Grade 7 Science - Pure Substances and Mixtures $\boldsymbol{Bill\ Nye-Atoms\ and\ Molecules}$ $\boldsymbol{Answers}$

1.	Things are made up of IIny pieces of stuff.
2.	"Atoms" is from a Greek word meaning <u>uncuttable</u> .
3.	The heavy particles of the atom are in the <u>middle</u> . The <u>light</u> particles are on the
	outside. Atoms are mostly <u>empty</u> <u>space</u>
4.	<u>Protons</u> and <u>Neutrons</u> are found in the nucleus. Their charges are <u>positive</u> and
	<u>neutral</u> , respectfully. <u>electrons</u> buzz around the nucleus at a <u>great/huge</u> distance away.
5.	Electrons <u>repel</u> each other and move around so fast that the atom acts like a <u>solid</u> .
6.	
7.	Atoms are the basic <u>building</u> <u>blocks</u> of all matter.
8.	When atoms combine, they can makemolecules
9.	All matter is made of <u>atoms</u> and <u>molecules</u>
10.	Water is _ <u>2</u> parts <u>hydrogen</u> and1 part <u>oxygen</u> . We know this because water can be
	split apart by <u>electrolysis</u>
11.	To "see" the spaces in matter, add one cup of water and one cup ofalcohol The total volume is
	not exactly two cups, since alcohol has a lot of empty space between molecules,
	where water molecules canfit
12.	The number ofprotons_ in the nucleus makes the atoms different. Elements have been arranged on a
	<u>periodic</u> <u>table</u> . The <u>atomic</u> number represents the number of protons.
13.	elements occur naturally.
14.	NaCl is SALT. C ₃ H ₅ N ₃ O ₉ is nitroglycerine and is used to makedynamite
15.	Inside protons and neutrons are <u>quarks</u> . These are studied by a <u>nuclear</u>
	<u>physicist</u>
16.	Everything in the universe is either <u>matter</u> or <u>energy</u> . The smallest pieces of matter are
	<u>atoms</u> .
17.	<u>carbon</u> is in every living thing, like us <u>Organic</u> chemistry is the study of this element's
	compounds.